The Patent Office
Concept House
Cardiff Road
Newport
South Wales
NP10 8QQ

I, the undersigned, being an officer duly authorised in accordance with Section 74(1) and (4) of the Deregulation & Contracting Out Act 1994, to sign and issue certificates on behalf of the Comptroller-General, hereby certify that annexed hereto is a true copy of the documents as originally filed in connection with the patent application identified therein.

In accordance with the Patents (Companies Re-registration) Rules 1982, if a company named in this certificate and any accompanying documents has re-registered under the Companies Act 1980 with the same name as that with which it was registered immediately before re-registration save for the substitution as, or inclusion as, the last part of the name of the words "public limited company" or their equivalents in Welsh, references to the name of the company in this certificate and any accompanying documents shall be treated as references to the name with which it is so registered.

In accordance with the rules, the words "public limited company" may be replaced by p.l.c., plc, P.L.C. or PLC.

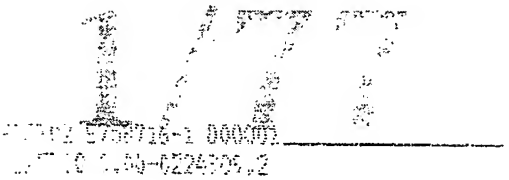
Re-registration under the Companies Act does not constitute a new legal entity but merely subjects the company to certain additional company law rules.

Signed

Dated 28 March 2006

Patents Form 1/77

Patents Act 1977
(Rule 16)



Request for grant of a patent

(See the notes on the back of this form. You can also get an explanatory leaflet from the Patent Office to help you fill in this form)

The Patent Office

Cardiff Road
Newport
South Wales
NP10 8QQ

1. Your reference 8459 GB PAB

2. Patent application number *(The Patent Office will fill in this part)* 0224909.2 25 OCT 2002

3. Full name, address and postcode of the or of each applicant *(underline all surnames)*

Norgine Europe B.V.
Hogehilweg 7
1101 CA Amsterdam ZO
Netherlands

Patents ADP number *(if you know it)* 7944671001

If the applicant is a corporate body, give the country/state of its incorporation: Netherlands

4. Title of the invention COLON CLEANSING COMPOSITIONS

5. Name of your agent *(if you have one)* Abel & Imray

"Address for service" in the United Kingdom to which all correspondence should be sent *(including the postcode)*

20 Red Lion Street
London
WC1R 4PQ
United Kingdom

Patents ADP number *(if you know it)* 174001

6. If you are declaring priority from one or more earlier patent applications, give the country and the date of filing of the or of each of these earlier applications and *(if you know it)* the or each application number

Country	Priority application number *(if you know it)*	Date of filing *(day / month / year)*

7. If this application is divided or otherwise derived from an earlier UK application, give the number and the filing date of the earlier application

Number of earlier application	Date of filing *(day / month / year)*

8. Is a statement of inventorship and of right to grant of a patent required in support of this request? *(Answer 'Yes' if:*

a) any applicant named in part 3 is not an inventor, or
b) there is an inventor who is not named as an applicant, or
c) any named applicant is a corporate body.
See note (d))

Yes

Patents Form 1/77

9. Enter the number of sheets for any of the following items you are filing with this form. Do not count copies of the same document

Continuation sheets of this form	-
Description	50
Claim(s)	9
Abstract	-
Drawing(s)	-

10. If you are also filing any of the following, state how many against each item.

Priority documents	-
Translations of priority documents	-
Statement of inventorship and right to grant of a patent *(Patents Form 7/77)*	-
Request for preliminary examination and search *(Patents Form 9/77)*	-
Request for substantive examination *(Patents Form 10/77)*	-
Any other documents *(please specify)*	-

11. I/We request the grant of a patent on the basis of this application.

Signature Abel and Imray Date 25-10-02

Abel & Imray

12. Name and daytime telephone number of person to contact in the United Kingdom

Paul Brady
020 7242 9984

Warning

After an application for a patent has been filed, the Comptroller of the Patent Office will consider whether publication or communication of the invention should be prohibited or restricted under Section 22 of the Patents Act 1977. You will be informed if it is necessary to prohibit or restrict your invention in this way. Furthermore, if you live in the United Kingdom, Section 23 of the Patents Act 1977 stops you from applying for a patent abroad without first getting written permission from the Patent Office unless an application has been filed at least 6 weeks beforehand in the United Kingdom for a patent for the same invention and either no direction prohibiting publication or communication has been given, or any such direction has been revoked.

Notes

a) If you need help to fill in this form or you have any questions, please contact the Patent Office on 08459 500505.

b) Write your answers in capital letters using black ink or you may type them.

c) If there is not enough space for all the relevant details on any part of this form, please continue on a separate sheet of paper and write "see continuation sheet" in the relevant part(s). Any continuation sheet should be attached to this form.

d) If you have answered 'Yes' Patents Form 7/77 will need to be filed.

e) Once you have filled in the form you must remember to sign and date it.

f) For details of the fee and ways to pay please contact the Patent Office.

Colon Cleansing Compositions

The present invention relates to orthostatic lavage solutions, colon evacuants or colon cleansing compositions, also known as lavage compositions for cleansing the gastrointestinal tract, and methods of use of such compositions.

General Background

Colon cleansing is important prior to numerous diagnostic and surgical procedures, for example before colonoscopy, barium enema examination or colon surgery. It is also useful for preventing infection after surgery on the lower intestine. Colon cleansing is also known as colon clearing.

A variety of methods for colon cleansing are known. Dietary manipulation, laxatives, cathartics and enemas were traditionally used (Thomas, G. *et al.*, *Gastroenterology*, 1982, **82**, 435-437). Sodium phosphate solutions (Clarkston, W.K. *et al.*, *Gastrointestinal Endoscopy*, 1996, **43**, 43-48) and magnesium citrate/sodium picosulphate solutions (Regev, A. *et al.*, *Am. J. Gastroenterol.*, 1998, **93**, 1478-1482) have also been used.

Those methods suffer from various drawbacks. Dietary manipulation and laxatives are time consuming; enemas are unpleasant for the patient; and dangerous salt and water losses may occur with cathartics, enemas and with sodium phosphate solutions.

Sodium phosphate solutions, such as that available from C.B. Fleet Company Inc. (4615 Murray Place, PO Box 11349, Lynchburg, Virginia 24506, USA) under the trade name Phospho-soda® are hyperosmotic solutions which increase retention of water in the intestine and thereby promote bowel movement. Phospho-soda comprises, per 5ml portion, 2.4g monobasic sodium phosphate monohydrate with 0.9g dibasic sodium

phosphate heptahydrate in a buffered aqueous solution. Typically 20 to 45 ml are taken by an adult patient followed by a large quantity of water. If the water is not taken, elevated serum sodium and phosphate levels may result, leading to serious kidney problems. The risk of those side effects makes it necessary for there to be direct medical supervision during administration of Phospho-soda.

Another approach to colon cleansing is orthostatic intestinal lavage, in which a large volume of an electrolyte solution is ingested, either by drinking or by infusion through a nasogastric tube. Such lavage solutions are also known as bowel lavage solutions. Consumption of the solution results in volume-induced diarrhoea and thus cleansing of the colon. The method is generally faster than the traditional approaches. The main component of early lavage solutions was sodium chloride. However, as a significant percentage of such saline-based lavage solutions is absorbed into the bloodstream in the gut of the patient, a rapid increase in intravascular volume results, which has caused serious complications in some patients.

In 1980, Davis and co-workers reported the development of a lavage solution, that they described as being associated with minimal water and electrolyte absorption or secretion (Davis G.R. *et al.*, Gastroenterology, 1980, **78**, 991-995). The solution included sodium sulphate and polyethylene glycol. Sulphate ions are poorly absorbed in the gut. As a result, sodium absorption is markedly reduced when sulphate, rather than chloride or bicarbonate, is the predominant counter-anion present in a lavage solution in the gut. In addition to sodium sulphate (40.0 mM, 5.68g/l), the solution described by Davis *et al.* comprises sodium chloride (25mM, 1.463g/l), potassium chloride (10mM, 0.745g/l), sodium bicarbonate (20mM, 1.680g/l), polyethylene glycol (PEG 4000 "carbowax", 64g/l) and water. The solution was administered

in a quantity of 4 litres. The solution was shown to be effective in cleansing the gastrointestinal tract and it has been commercialised under the trade name GoLYTELY® (Braintree Laboratories Inc, Braintree, Massachusetts, U.S.A.). The commercially available GoLYTELY composition, as available after August 1996 and at the time of filing, is supplied in dry powder form comprising sodium sulphate (40.0 mM, 5.685g/l), sodium chloride (25mM, 1.464g/l), potassium chloride (10mM, 0.743g/l), sodium bicarbonate (20mM, 1.685g/l) and PEG 3350 polyethylene glycol (59g/l) for making up to 4 litres. GoLYTELY is also supplied in aqueous solution.

The GoLYTELY solution, whilst effective, has a very salty taste, which adversely affects patient compliance. Typically the composition is presented as four or more litres of aqueous solution, and it is important that the whole prescribed volume is consumed. Consumption of such large volumes of fluid can also affect compliance adversely.

Fordtran *et al.* (WO87/00754) subsequently developed a reduced sodium sulphate solution (RSS) comprising no sodium sulphate but instead having a relatively high concentration of polyethylene glycol (75 to 300g/l). The preferred solution disclosed in WO87/00754 comprises PEG 3350 (120g/l), sodium bicarbonate (1.68g/l), potassium chloride (0.74g/l) and sodium chloride (1.46g/l) and it is also administered in a quantity of 4 litres. A solution very similar to the preferred solution of WO87/00754 is commercialised by Braintree Laboratories Inc (Braintree, Massachusetts, U.S.A.) under the name NuLYTELY® (initially also under the name GoLYTELY-RSS). The NuLYTELY composition comprises PEG 3350 (105g/l), sodium bicarbonate (1.43g/l), potassium chloride (0.37g/l) and sodium chloride (2.80g/l) and it is supplied in dry powder form for making up to 4 litres.

Whilst being effective in colon cleansing in the clinic, both the GoLYTELY and the NuLYTELY solutions must be ingested in large quantities, typically four litres. Ingestion of such volumes of gut lavage solution is generally physically unpleasant or even impossible for many patients, may result in retching, and is time consuming. In spite of the absence of sodium sulphate in NuLYTELY, both NuLYTELY and GoLYTELY have an unpleasant salty taste. The unpleasant taste exacerbates the problem of patient compliance, particularly when the patient is not under medical supervision.

In WO 89/05659 (Borody) there is described an orthostatic lavage solution comprising polyethylene glycol, electrolytes and from 0.25 to 50g/l ascorbic acid (vitamin C) or a salt thereof. The presence of ascorbic acid or a salt thereof is said to reduce the required volume of solution to 3 litres or less. Whilst about 3g of ascorbic acid may be absorbed in the intestine (Hornig, D. *et al.*, Int. J. Vit. Nutr. Res., 1980, **50**, 309) any further ascorbic acid is reported in WO 89/05659 to contribute to the diarrhoea and to inhibit bacterial gas generation and bacterial reproduction. The ascorbic acid is also said to facilitate ingestion of the lavage solution because its pleasant acidic taste masks the usual nauseating taste of the salty polyethylene glycol solution.

The solutions described by Borody comprise polyethylene glycol (preferably PEG 3350 or PEG 4000) at a concentration of 30-60 g/litre together with inorganic electrolytes (sodium chloride, potassium chloride, sodium hydrogen carbonate and sodium sulfate). In any given solution, the quantity of PEG is described as being adjusted such that the osmolarity of the solution is approximately 289 mOsmol/kg (i.e. isotonic). The osmolarity of a solution may be measured using conventional laboratory techniques. It is also possible to calculate osmolarity from a knowledge of the components of a

solution. Details of osmolarity calculations are given herein below.

A formulation as described by Borody has been available on the market in Australia for more than 10 years under the tradename GLYCOPREP C (Pharmatel). The GLYCOPREP C dry composition comprises PEG 3350 (53g/l), sodium chloride (2.63g/l), potassium chloride (0.743g/l), sodium sulphate (5.6g/l), ascorbic acid (6g/l), aspartame (0.360g/l), citric acid (0.900g/l) and lemon flavour (0.090g/l). 3 litres of the solution are generally administered.

Whilst the addition of ascorbic acid goes some way towards providing an improved bowel preparation, that preparation must be ingested in quantities of approximately 3 litres. Ingestion of such volumes of gut lavage solution is still generally physically unpleasant or, for some patients, even impossible, may result in retching, and is time consuming. Accordingly there remains a requirement for lavage solutions with a more pleasant taste that are effective in a smaller volume.

Colon clearance is important before numerous surgical or diagnostic procedures, including colonoscopy, barium enema examination, sigmoidoscopy and colon surgery. It is desirable that the colon clearance may be carried out by the patient himself or herself without medical supervision at home in advance of attending the hospital or surgery where the surgical or diagnostic procedure is to take place. It is important that patient compliance is good without medical supervision if satisfactory colon clearance is to be achieved.

The compositions of the prior art are summarised in Table 1. In that table, the indicated quantities are the quantities present per litre of aqueous solution. The calculated osmolarity of the solutions (in mOsmol/kg) is also

given in the table together with the recommended dose (in litres).

Table 1: Composition of prior art colon cleansing treatments

Formulation	PEG g	Na_2SO_4 g	$NaHCO_3$ g	NaCl g	KCl g	Vit C g	Osm	Vol l
GoLYTELY	60	5.7	1.93	1.46	0.75	-	255	4
NuLYTELY	105	-	1.43	2.8	0.37	-	176	4
Glycoprep C	53	5.6	-	2.63	0.74	6.0	291	3

Description of the invention

It has now been found, surprisingly, that a cleansing solution comprising an alkali metal or alkaline earth metal sulphate, ascorbic acid and/or one or more salts thereof, a relatively high concentration of PEG and, optionally, further electrolytes, has a cleansing action that is effective when administered in a small volume, and is palatable. The cleansing solution comprising a composition of the invention achieves satisfactory colon cleansing when used in a quantity of approximately 2 litres. Conventional cleansing solutions must be used in a quantity of at least 3 to 4 litres.

PEG has been known to contribute towards the diarrhoea-producing effect of PEG-containing solutions by promoting malabsorption of electrolytes. However, it has now been found, surprisingly, that a cleansing solution which comprises an alkali metal or alkaline earth metal sulphate, ascorbic acid and/or one or more salts thereof, a relatively high concentration of PEG and, optionally, further electrolytes, has a powerful cleansing or purging action. Hence it has been found that smaller volumes of solution are needed and yet the solution remains palatable. The cleansing solution achieves satisfactory colon cleansing for, e.g. colonoscopy, when used in a quantity of approximately 2 litres.

The invention provides a dry composition for admixture with water wherein the dry composition comprises, per litre of aqueous solution to be made, the following components:

a) 80 to 350g of a polyethylene glycol;

b) 3 to 20g of ascorbic acid, one or more salts of ascorbic acid or a mixture of ascorbic acid and one or more salts of ascorbic acid;

c) 1 to 15g of an alkali metal or alkaline earth metal sulphate or a mixture of alkali metal or alkaline earth metal sulphates; and

d) optionally one or more electrolytes selected from sodium chloride, potassium chloride and sodium hydrogen carbonate;

the components of the composition being selected such that an aqueous solution made up to 1 litre has an osmolarity within the range of from 300 to 700 mOsmol/kg.

The invention also provides a cleansing solution comprising an aqueous solution of the dry composition of the invention, the components having the concentrations stated above, the composition having an osmolarity within the range defined above, and the volume of the composition being from 0.5l to 5l.

The solutions of the invention are not isotonic i.e. they do not have the same osmotic pressure as the blood in the gut vasculature. The solutions are, however, approximately iso-osmolar, that is to say, the solution excreted from the patient of a solution has substantially the same ion content as the solution ingested. Consequently, there is no substantial net change in the ion levels in the blood of the patient.

The osmolarity of a solution is the number of non-permeating particles dissolved in a solution. For a substance that remains completely associated as a unit in solution (e.g. a neutral organic molecule) the osmolarity and

the molarity of a solution are essentially the same. For a substance that dissociates when it dissolves (e.g. an ionic salt), the osmolarity is the number of moles of individual dissolved species in solution after dissolution.

The osmolarity of a solution can be measured using standard laboratory techniques. It can also be calculated from a knowledge of the components of a solution. As an example, the osmolarity of the GoLytely solution may be calculated as follows:

PEG: 60g, MW=3350, one species per mole in solution:
Contribution = 60/3350*1 = 18.0 mOsmol/kg

Na_2SO_4: 5.7g, MW=142, three species per mole in solution:
Contribution = 5.7/142*3 = 120.4 mOsmol/kg

$NaHCO_3$: 1.93g, MW=84, two species per mole in solution:
Contribution = 1.93/84*2 = 46.0 mOsmol/kg

NaCl: 1.46g, MW=58.5, two species per mole in solution:
Contribution = 1.46/58.5*2 = 50.0 mOsmol/kg

KCl: 0.75g, MW= 74.5, two species per mole in solution:
Contribution = 0.75/74.5*2 = 20.1 mOsmol

Total Osmolarity = 255 mOsmol/kg

In some cases, a calculated osmolarity disagrees with a measured osmolarity. There are a variety of possible reasons for that, mostly connected with the fact that the number of free dissolved species in solution may not be exactly that assumed from ideal behaviour. For example, if several components are present, those may aggregate and lead to the number of independent dissolved species being lower than that calculated. As a further example, in dependence on the pH of the solution, organic acids and bases can be incompletely dissociated or associated.

A cleansing solution comprising PEG at a concentration of over 100g/l has been described previously (NuLYTELY). It has generally been assumed that cleansing solutions must be isotonic, i.e. have the same osmolarity as the vascular fluid in the gut. The high concentration of PEG was thus accompanied by a low concentration of electrolyte salts so that the cleansing solution was isotonic. For example, sodium sulfate is omitted from the NuLYTELY solution. It has now been found, surprisingly, that it is not necessary for the cleansing solution to be isotonic and, furthermore, that a hypertonic solution comprising PEG, an alkali metal or alkaline earth metal sulphate or a mixture of alkali metal or alkaline earth metal sulphates, electrolytes and ascorbic acid and/or one or more salts thereof is a cleansing solution that is more effective than isotonic solutions of the prior art.

In healthy volunteers, at an administered volume of 2 litres, a hypertonic cleansing solution of the present invention has been found to cause a 50% increase in stool weight and stool volume output compared with an isotonic solution lacking sodium sulphate and ascorbic acid but otherwise having the same composition, that is to say, the same concentrations of PEG, sodium bicarbonate, sodium chloride and potassium chloride. No adverse side effects were observed. The hypertonic cleansing solution of the invention was also found to be more effective at that administered volume than prior art compositions that are isotonic and comprise a lower concentration of sodium sulphate.

Preferably the osmolarity of a cleansing solution of the present invention is 330 mOsmol/kg or greater, more preferably 350 mOsmol/kg or greater, still more preferably 400 mOsmol/kg or greater, for example 460 mOsmol/kg or greater. Preferably, the osmolarity of the cleansing

solution of the present invention is 600 mOsmol/kg or lower, more preferably 550 mOsmol/kg or lower, still more preferably 500 mOsmol/kg or lower, for example 470 mOsmol/kg or lower. For example the osmolarity may be in a range wherein the lower limit is selected from any of 330, 350, 400 and 460 mOsmol/kg, and the upper limit is selected, independently, from any of 600, 550, 500 and 470 mOsmol/kg.

Whereas previously it was thought to be necessary for a cleansing solution to be isosmolar, and pains were taken to adjust them to be so, it has now, surprisingly, been found that high osmolarity is not only safe, but more effective than prior art solutions and that patients are less likely to vomit with the lower volume of ingested fluid. When the osmolarity is contributed to by the PEG, the double effect of high PEG concentration and increased osmolarity drives the cleansing solution at a higher pace with reduced side effects and yet with greater safety. From the resulting effluent volume measurements we have found that the combination of the two effects is synergistic.

The polyethylene glycol (PEG) used in a composition of the present invention preferably has an average molecular weight of 2000 or greater. Preferably the PEG has an average molecular weight of 2500 or greater. Preferably the PEG has an average molecular weight of 4500 or lower. For example the PEG may be PEG 3350 or PEG 4000. Optionally, the PEG used in a composition of the invention may comprise two or more different PEG species. A composition of the invention preferably comprises 90g or more of PEG per litre, more preferably 100g or more of PEG per litre. Preferably, a composition of the invention comprises 250g or less of PEG per litre, more preferably 150g or less of PEG per litre, still more preferably 140g or less of PEG per litre, still more preferably 125g or less of PEG per litre. For example, a composition of the present invention may comprises PEG at a

concentration within a range wherein the lower limit is 90 or 100g per litre and the upper limit is, independently, 350, 250, 150 or 125g per litre. For example, a composition of the invention may comprise 100 or 125g per litre. Most preferably a composition of the invention comprises 100g of PEG per litre.

Preferably the alkali metal or alkaline earth metal sulphate or the mixture of alkali metal or alkaline earth metal sulphates is present in a cleansing composition of the invention in a quantity of 2g or more per litre, more preferably in a quantity of 3g or more per litre, still more preferably in a quantity of 5g or more per litre. Preferably the alkali metal or alkaline earth metal sulphate or the mixture of alkali metal or alkaline earth metal sulphates is present in the cleansing compositions of the invention in a quantity of 10g or less per litre, more preferably in a quantity of 9g or less per litre, still more preferably in a quantity of 7.5g or less per litre. For example, the alkali metal or alkaline earth metal sulphate or the mixture of alkali metal or alkaline earth metal sulphates may be present in a quantity within a range in which the lower limit is selected from any of 2, 3 and 5 g per litre and the upper limit is selected, independently, from any of 10, 9 and 7.5g per litre. For example the alkali metal or alkaline earth metal sulphate or the mixture of alkali metal or alkaline earth metal sulphates is present in a quantity of 5g or 7.5g per litre, most preferably 7.5g per litre.

The alkali earth metal or alkaline earth metal may be, for example, sodium, magnesium or calcium. Sodium is generally preferred, but magnesium or calcium may be used.

A composition of the invention preferably comprises sodium chloride. Sodium chloride is preferably present in a quantity of 0.5g or more per litre, more preferably 1g or more per litre, still more preferably a quantity of 2g or

more per litre. Sodium chloride is preferably present in a quantity of 7g or less per litre, more preferably 5g or less per litre, still more preferably a quantity of 4g or less per litre. For example, sodium chloride may be present in at a concentration within a range in which the lower limit is selected from any of 0.5, 1 and 2 g per litre and the upper limit is selected, independently, from any of 7, 5 and 4 g per litre.

A composition of the invention preferably comprises potassium chloride. Preferably potassium chloride is present in a quantity of 0.2g or more per litre, more preferably in a quantity of 0.5g or more per litre, most preferably in a quantity of 0.7g or more per litre. Preferably potassium chloride is present in a quantity of 4g or less per litre, more preferably in a quantity of 2g or less per litre, most preferably in a quantity of 1.3g or less per litre. For example, potassium chloride may be present in at a concentration within a range in which the lower limit is selected from any of 0.2, 0.5 and 0.7g per litre and the upper limit is selected, independently, from any of 4, 2, and 1.3 g per litre.

A composition of the invention may comprise sodium bicarbonate. Because of the reaction between sodium bicarbonate and acids, bicarbonate ions are generally destroyed, with accompanying effervescence as CO_2 is produced, on addition of water to a composition comprising ascorbic acid and a bicarbonate. The same reaction may occur in a dry powder composition if small amounts of moisture, for example atmospheric moisture, are present. The reaction between bicarbonate and ascorbic acid in the dry powder composition may be avoided if coated ascorbic acid is used. The reaction may also be avoided by packaging the dry composition in two separate individual units such that the bicarbonate and the ascorbic acid are not in contact.

The term "ascorbate component" is used herein to denote the ascorbic acid, one or more salts thereof or a mixture of ascorbic acid that is used in a composition of the present invention. The ascorbate component is present in a composition of the invention in a quantity of from 3-20g per litre of solution. Preferably the ascorbate component is present in a quantity of 4g or more per litre, more preferably in a quantity of 5g or more per litre. Preferably the ascorbate component is present in a quantity of 15g or less per litre, more preferably in a quantity of 10g or less per litre. For example, the ascorbate component may be present in a quantity within a range in which the lower limit is 4 or 5 g per litre and the upper limit is, independently, 15 or 10g per litre. For example, the ascorbate component is present in a quantity of 5 to 10g per litre, for example, 5 or 10 g per litre.

Preferred salts of ascorbic acid are alkali metal and alkaline earth metal salts, for example sodium ascorbate, potassium ascorbate, magnesium ascorbate and calcium ascorbate. A particularly preferred salt of ascorbic acid is sodium ascorbate. Preferably the ascorbate component comprises both ascorbic acid and one or more salts thereof. Preferably the ascorbic acid and the salt(s) thereof are present in a weight ratio within the range of from 1:9 to 9:1. Ascorbic acid and salts thereof may, in practice, be provided as hydrates. If a hydrate is used, the weight and/or weight ratio mentioned here is the weight and/or weight ratio of ascorbic acid or salt(s) thereof without water of hydration. Preferably the ascorbic acid and the salt(s) thereof are present in a weight ratio within the range of from 2:8 to 8:2, more preferably 3:7 to 7:3, still more preferably 4:6 to 6:4, for example 4.7 to 5.9.

It has been found previously by others that the plasma bicarbonate ion level may fall following use of cleansing

solutions based on 0.9% saline or 7.2% mannitol that do not contain a balanced amount of bicarbonate. A lowered plasma bicarbonate level may have serious adverse clinical consequences associated with a reduced blood pH (acidosis) and consequent reduced capacity to transport CO_2 in the bloodstream. Acidosis may lead to weakness, disorientation, coma and eventually death. However, it has now been found according to the present invention that plasma bicarbonate lowering is much reduced by the use of a composition comprising both ascorbic acid and one or more salts thereof. The presence of an ascorbate salt contributes to the osmotic load of the solution and also aids the maintenance of the bicarbonate level. This is a further advantage of the compositions of the present invention.

Compositions of the invention are preferably flavoured. Flavouring for use in compositions of the invention should preferably mask saltiness, be relatively sweet but not excessively so, and be stable in the composition. Flavouring makes the solutions more palatable and thus aids patient compliance. Preferred flavourings include lemon e.g. Ungerer Lemon (available from Ungerer Limited, Sealand Road, Chester, England CH1 4LP) strawberry e.g. Ungerer Strawberry, grapefruit e.g. Ungerer Grapefruit flavouring powder, blackcurrant e.g. Ungerer Blackcurrant, pineapple e.g. IFF (International Flavours and Fragrances) Pineapple flavouring powder and vanilla/lemon and lime e.g. IFF Vanilla and Givaudin Roure Lemon and Lime Flav-o-lok. Those and further suitable flavourings are available from International Flavors and Fragrances Inc. (Duddery Hill, Haverhill, Suffolk, CB9 8LG, England), Ungerer & Company (Sealand Road, Chester, England CH1 4LP) or Firmenich (Firmenich UK Ltd., Hayes Road, Southall, Middlesex UB2 5NN). More preferred flavourings are lemon, kiwi, strawberry and grapefruit. The most preferred flavouring is lemon.

Preferably compositions of the invention comprise a sweetener. Sugar-based sweeteners are not suitable because delivery of unabsorbed sugars to the colon provides a substrate for bacteria. Such sugars may be metabolised by the bacteria to form explosive gases such as hydrogen and methane. The presence of explosive gases in the colon can be highly dangerous when electrical apparatus is to be used during colonoscopy or other procedures. Preferred sweeteners include aspartame, acesulfame K and saccharine or combinations thereof. Citric acid may also be present as a taste enhancer.

The ascorbic acid and/or salt(s) of ascorbic acid in a dry composition of the present invention may be coated. A coating helps to maintain the stability of the ascorbic acid and/or the salt(s) thereof. As stated above, ascorbic acid and salts thereof are otherwise poorly stable in the presence of moisture.

A dry composition in accordance with the invention may be in powder, granular or any other suitable physical form. A dry composition of the invention may be provided in unit dosage form, for example, in a sachet. Preferably a dry composition is provided in two or more component form, in which the ascorbic acid and/or salt(s) thereof are packaged separately from other components. For example, a first component, for example, in a unit dose form, for example, a sachet may contain polyethylene glycol, sodium sulphate, sodium chloride, potassium chloride, sweetening and flavouring agents, and a second component, for example, a unit dose form, for example, a sachet containing ascorbic acid and sodium ascorbate.

A composition of the invention may be provided as a solution in water, for example, in one or more containers, each containing, for example, 0.5 or 1 litre of solution.

The present invention also provides a method of cleansing the colon of a mammal, comprising administering orally to the mammal a cleansing fluid comprising, per litre, the following components:

a) 80 to 350g of a polyethylene glycol;

b) 3 to 20g of ascorbic acid, one or more salts of ascorbic acid or a mixture of ascorbic acid and one or more salts of ascorbic acid;

c) 1 to 15g of an alkali metal or alkaline earth metal sulphate or a mixture of alkali metal or alkaline earth metal sulphates; and

d) optionally one or more electrolytes selected from sodium chloride, potassium chloride and sodium hydrogen carbonate;

the components of the composition being selected such that the cleansing fluid has an osmolarity within the range of from 300 to 700 mOsmol/kg, the volume of fluid administered being from 1.5 to 3 litres for an adult human and pro rata for a mammal other than an adult human.

The exact quantity of the solution of the invention to be administered will depend on the patient being treated. For example, a smaller volume of cleansing solution is appropriate in the treatment of small children and a higher volume of cleansing solution is appropriate in patients with prolonged colonic transit times.

The method of the present invention may be used to cleanse the colon prior to carrying out a diagnostic, therapeutic or surgical procedure on the colon, rectum or anus or elsewhere in the abdomen. The diagnostic or surgical procedure may, for example, be colonoscopy, barium enema examination, sigmoidoscopy or colon surgery.

The method of the present invention may also be used in the treatment of acute gastrointestinal infections, for example bacterial or viral gastroenteritis. The aim in such

a treatment is to remove stools from the infected colon so that the patient absorbs fewer toxins and has a shorter period of diarrhoea, toxicity, anorexia, nausea or vomiting. Upon developing diarrhoea, cramping and malaise the use of the purgative product removes from the bowel the offending, infected column of bowel flora, so ameliorating the infection in a short length of time.

Preferably, the total volume of solution is administered over 1 to 4 hours. The 1 to 4 hours may be in a continuous period or a discontinuous period. In discontinuous administration, a portion of the solution, typically approximately half, may be administered the evening before the diagnostic, therapeutic or surgical procedure is to be carried out, with the remainder of the solution being administered on the day of the procedure.

A composition for use in a method of the invention has the preferred features described above in respect of the composition of the invention.

The invention further provides a dry composition for admixture with water wherein the dry composition comprises, per litre of aqueous solution to be made, the following components:

- 30 to 350 g of a polyethylene glycol
- 3 to 20g of ascorbic acid and one or more salts of ascorbic acid
- optionally one or more electrolytes selected from sodium chloride, potassium chloride, sodium hydrogen carbonate and the alkali metal or alkaline earth metal sulphates.

The invention also provides a solution of the above dry composition.

It has been found, surprisingly, that a colon cleansing solution comprising ascorbic acid and one or more salts of ascorbic acid has fewer side effects than a cleansing solution comprising ascorbic acid and no salts thereof.

Furthermore, a colon cleansing solution comprising ascorbic acid and one or more salts of ascorbic acid has been found to be even more efficacious in its colon cleansing action than a solution comprising ascorbic acid and no salts thereof. Cleansing solutions comprising a salt of ascorbic acid but no ascorbic acid have also been found to be less efficacious than solutions comprising both ascorbic acid and one or more salts thereof.

The levels of plasma bicarbonate and other anion may fall during use of cleansing solutions comprising ascorbic acid alone. The presence of one or more salts of ascorbic acid contribute to the osmotic load of the solution and also aid the maintenance of the plasma bicarbonate level. The fall in the level of plasma bicarbonate is much reduced by the use of a composition comprising both ascorbic acid and one or more salts thereof.

Any suitable salt of ascorbic acid may be used. Preferred salts of ascorbic acid are alkali metal and alkaline earth metal salts, for example sodium ascorbate, potassium ascorbate, magnesium ascorbate and calcium ascorbate. A particularly preferred salt of ascorbic acid is sodium ascorbate. Preferably, the salt is sodium ascorbate.

Preferably the ascorbic acid and the salt(s) thereof are present in a weight ratio in the range of from 1:9 to 9:1. Ascorbic acid or salts thereof may, in practice, be provided as hydrates. If a hydrate is used, the weight and/or weight ratio mentioned herein is the weight and/or weight ratio of the ascorbic acid and the salt thereof without water of hydration. Preferably the ascorbic acid and the salt thereof are present in a weight ratio in the range of 2:8 to 8:2, more preferably 3:7 to 7:3, still more preferably 4:6 to 6:4, for example 4.7:5.9.

Preferably, a composition of the invention comprising ascorbic acid and one or more salts thereof further comprises

one or more electrolytes selected from sodium chloride, potassium chloride, sodium hydrogen carbonate and sodium sulphate. Preferably, a composition of the invention comprises sodium sulphate.

A composition of the invention as described above may be provided as a solution in water or as a dry composition for making up into a solution. In such a dry formulation, the ascorbic acid and/or the salt(s) of ascorbic acid may be coated. Such a coating helps to maintain stability of ascorbic acid or salt(s) thereof. Ascorbic acid and salts thereof are otherwise poorly stable in the presence of moisture.

A dry composition may be provided in a unit dosage form, for example, in a sachet. A dry composition may be provided in two or more component form, in which the ascorbic acid and/or the salts thereof are packaged separately from other components. For example, a first unit dosage form, for example, a first sachet may contain polyethylene glycol, sodium sulphate, sodium chloride, potassium chloride, sweetening and flavouring and a second unit dosage form, for example, a second sachet may contain ascorbic acid and sodium ascorbate.

The present invention further provides a method of cleansing the colon of a mammal, comprising administering orally to the mammal a preparation comprising, per litre the following components:

- 30 to 350 g polyethylene glycol
- 3 to 20 g of a mixture of ascorbic acid and a salt of ascorbic acid
- optionally one or more electrolytes selected from sodium chloride, potassium chloride, sodium hydrogen carbonate and the alkali metal or alkaline earth metal sulphates,

the volume of composition administered being from 1.5 to 4 litres for an adult human and pro rata for a mammal other than an adult human. The exact quantity of solution to be administered will depend on the patient being treated. For example, a smaller dose of cleansing solution is appropriate in the treatment of small children and a higher dose of cleansing solution is appropriate in patients with prolonged colonic transit times.

A method of the present invention may be used to cleanse the colon prior to carrying out a diagnostic, therapeutic or surgical procedure on the colon, rectum or anus or elsewhere in the abdomen. A diagnostic or surgical procedure may, for example, be colonoscopy, barium enema examination, sigmoidoscopy or colon surgery. The method of the present invention may also be used in the treatment of acute gastrointestinal infections, for example bacterial or viral gastroenteritis.

Preferably, the total volume of fluid is administered over 1 to 4 hours. The 1 to 4 hours may be in a continuous period or may be in a discontinuous period. In one mode of administration, a portion of the solution, typically approximately half, may be administered the evening before the diagnostic, therapeutic or surgical procedure is to be carried out, with the remainder of the solution being administered on the day of the procedure.

An ascorbic acid and ascorbic acid salt-containing preparation for use in a method of the invention has the preferred features described above in respect of the corresponding ascorbic acid and ascorbic acid salt-containing preparation of the invention.

The present invention further provides the use of a PEG for the manufacture of a medicament for cleansing the colon of a mammal according to an administration regime comprising the consecutive steps of

a) administering 0.5 to 3.0 litres of a colon cleansing solution comprising a PEG (volume V_{PEG}) over a period of time t_1 and

b) administering 0.3 to 2.0 litres of clear fluid (volume V_{cf}) over a period of time t_2.

The invention also provides a method of cleansing the colon of a mammal, comprising administering orally to the mammal in consecutive steps:

a) 0.5 up to 3.0 litres of a cleansing solution comprising PEG (volume V_{PEG}) over a period of time t_1 and

b) 0.3 up to 2.0 litres of clear fluid (volume V_{cf}) over a period of time t_2.

t_1 is preferably up to 2 hours, more preferably up to 1 hour 30 minutes, typically approximately one hour. t_1 is preferably greater than 15 minutes, more preferably greater than 30 minutes. Similarly, t_2 is preferably up to 2 hours, more preferably up to 1 hour 30 minutes, typically approximately one hour. t_2 is preferably greater than 15 minutes, more preferably greater than 30 minutes.

Preferably, V_{PEG} is 500ml or greater, more preferably, V_{PEG} is 800ml or greater. Preferably, V_{PEG} is 2000ml or less, more preferably, V_{PEG} is 1500ml or less. For example V_{PEG} is approximately 1000ml. Preferably, V_{cf} is 300ml or greater, more preferably, 400ml or greater. Preferably, V_{cf} is 1500ml or less, preferably 1000ml or less. For example V_{cf} is approximately 500ml. In practice, under supervision in the clinic, clear fluid may be given until faecal output is clear and no longer contains any solid material.

Using the regime of the invention, it is found, surprisingly, that the weight of stool recovered is increased in comparison to a normal cleansing regime, in which only colon cleansing solution is administered. The invention hails the start of a new era in which use of hypertonic solutions comprising PEG, together with added electrolytes

means that patients will have to drink added water to provide the orthostatic lavage power. The markedly reduced volume of the active solution required to be drunk may be followed by any fluid that the patient chooses, including water, lemonade and others.

It is postulated by the present inventors that the total ingested osmotic load is of importance in determining the success of the colon cleansing action. Using the regime of the invention, ingestion of the required osmotic load is possible in a shorter time period which makes the onset of the effect of the cleansing more rapid.

The use of a clear fluid enables the progress of the colon cleansing, including the end point, to be assessed by visual inspection of the faecal output. When the faecal output is clear, no further fluid need be ingested by the patient. The clear fluid may be any fluid that allows inspection of colonic output. Typically the clear fluid is a water-based beverage, including, for example, water, lemonade, cola drinks, cordial drinks, clear fruit juices and even clear alcohol-containing beverages, for example beer. It is desirable that the clear fluid does not contain substantial amounts of or essentially any dietary fibre, as such fibre interferes with the cleansing of the colon according to the present invention. Accordingly, fruit juices, for example orange juice and kiwi juice, and fruit "squashes" should be strained before use. Clear fruit cordials, for example, lime cordial, are generally suitable. In view of the desirability of avoiding drinks containing glucose, so as to reduce the risk of explosive concentrations of hydrogen or methane building up in the gut, "diet" drinks containing no or low sugar are especially suitable, for example liquid drinks for diabetics, diet Coke (RTM), diet lemonade, dietary carbonated drinks or dietary cordials.

In general, the larger the volume of cleansing solution that is administered, the greater the quantity of stools that is collected. As mentioned in the introduction in relation to the prior art, 4 litres of colon cleansing solution is generally administered over 3 to 4 hours. Colon cleansing solutions generally have an unpleasant taste and many patients have difficulty ingesting the large quantity of solution typically necessary. It has now been found according to a further aspect of the present invention that highly effective colon cleansing can be achieved by administering first a cleansing solution, the volume being less than the volume described in the prior art, followed by administration of a clear fluid.

Suitable colon cleansing solutions for use in the method include in particular the colon cleansing solutions of the present invention described above.

Patient compliance is improved because the volume of cleansing solution that must be ingested is smaller than in the prior art methods. In comparison with ingestion of a cleansing solution of volume (V_{PEG}+ V_{cf}) but with the same total quantity of composition components, the effectiveness is, surprisingly, not reduced. Water alone is not active as a cleansing solution. It is normally simply absorbed in the gut.

Preferably, the colon is cleared prior to carrying out a diagnostic, therapeutic or surgical procedure on the colon, rectum or anus or elsewhere in the abdomen. The diagnostic or surgical procedure may, for example, be colonoscopy, barium enema examination, sigmoidoscopy or colon surgery.

As a variant of the two step aspect of the invention, there is provided the use of a PEG for the manufacture of a medicament for cleansing the colon of a patient according to an administration regime comprising the consecutive steps of:

a) administering 0.5 up to 3.0 litres of a first PEG-containing colon cleansing solution (volume V_{PEG}) over a period of time t_1

b) administering 0.3 up to 2.0 litres of clear fluid (volume V_{cf}) over a period of time t_2, and

c) administering 0.5 up to 3.0 litres of a second PEG-containing colon cleansing solution (volume $V2_{PEG}$) over a period of time t_3.

The present invention also provides a method for cleansing the colon of a patient according to an administration regime comprising the consecutive steps of:

a) administering 0.5 up to 3.0 litres of a first PEG-containing colon cleansing solution (volume V_{PEG}) over a period of time t_1

b) administering 0.3 up to 2.0 litres of clear fluid (volume V_{cf}) over a period of time t_2, and

c) administering 0.5 up to 3.0 litres of a second PEG-containing colon cleansing solution (volume $V2_{PEG}$) over a period of time t_3.

It is found that the effectiveness and the patient compliance is further increased when the colon cleansing solution is administered in two doses separated by ingestion of a volume of water, when compared with the administration of a single dose of total equal volume (i.e. $V_{PEG}+V2_{PEG}$).

Preferably, V_{PEG} is 500ml or greater, more preferably, V_{PEG} is 800ml or greater. Preferably, V_{PEG} is 2000ml or less, more preferably, V_{PEG} is 1500ml or less. For example V_{PEG} is approximately 1000ml. Preferably, $V2_{PEG}$ is 500ml or greater, more preferably, $V2_{PEG}$ is 800ml or greater. Preferably, $V2_{PEG}$ is 2000ml or less, more preferably, $V2_{PEG}$ is 1500ml or less. For example $V2_{PEG}$ is approximately 1000ml. Preferably, V_{cf} is 300ml or greater, more preferably, 400ml or greater. Preferably, V_{cf} is 1500ml or less, preferably 1000ml or less. For example V_{cf} is approximately 500ml.

t_1 is preferably 15 minutes or greater, more preferably from 30 minutes or greater. t_1 is preferably 2 hours or less, more preferably 1 hour 30 minutes or less. Typically t_1 is approximately one hour. Similarly, t_2 is preferably 15 minutes or greater, more preferably from 30 minutes or greater. t_2 is preferably 2 hours or less, more preferably 1 hour 30 minutes or less. Typically t_2 is approximately one hour. t_3 is preferably 15 minutes or greater, more preferably 30 minutes or greater. t_3 is preferably 2 hours or less, more preferably 1 hour 30 minutes or less. Typically t_3 is approximately one hour.

Preferably, the administration of the second dose of colon cleansing solution is followed by a second dose of clear fluid ($V2_{cf}$) over a period of time t_4. Preferably, $V2_{cf}$ is 500ml or greater, more preferably, 800ml or greater. Preferably, $V2_{cf}$ is 2000ml or less, preferably 1500ml or less. For example $V2_{cf}$ is approximately 1000ml. In practice, under supervision in the clinic, clear fluid may be given until faecal output is clear and no longer contains any solid material. t_4 is preferably 30 minutes or greater, more preferably 1 hour or greater. t_4 is preferably 3 hours or less, more preferably 2 hour 30 minutes or less. Typically t_4 is approximately two hours.

The volume of stools produced is significantly increased by the addition of steps in which clear fluid is administered in accordance with the invention. The acceptability of the treatment to the patient is much increased. Of nine volunteer subjects, eight preferred administration of the cleansing solution in two doses separated by a dose of water over administration of the cleansing solution in a single, larger dose.

Suitable colon cleansing compositions for use in the method of the invention include in particular the compositions of the invention described above.

A composition may be provided in two or more component form. For example, a first component may be a composition for making up a first PEG-containing colon cleansing, a second component being a composition for making up a second PEG-containing colon cleansing solution. Preferably, one or both of the components comprise(s) ascorbic acid and/or a salt thereof. The two components are preferably in unit dosage form, for example, comprising the composition in a sachet or other appropriate container. In such an arrangement, the ascorbic acid and/or the salts thereof are preferably packaged separately from other components. For example a first sachet may contain polyethylene glycol, sodium sulphate, sodium chloride, potassium chloride, sweetening and flavouring and a second sachet may contain ascorbic acid and sodium ascorbate, those sachets together being for making up the first colon cleansing solution. The a third sachet maybe provided, containing polyethylene glycol, sodium sulphate, sodium chloride, potassium chloride, sweetening and flavouring and a fourth sachet containing ascorbic acid and sodium ascorbate, those sachets together being for making up the second colon cleansing solution.

The various two or more component systems for providing compositions of the invention generally comprise the relevant composition in unit dosage form. A unit dose is generally an amount of dry composition suitable for making up to a defined volume with water. The volume may be any suitable volume, for example, for use in a two step or multi-step regime as described above, each unit dose may be suitable for making the total volume of solution for use in one of the defined cleansing steps. Alternatively, a unit dose may be suitable for making up to a defined volume, for example, a litre of cleansing solution.

It is convenient for the patient to provide the dry composition in the form of a kit, for example, a box, comprising the composition and instructions for its use. The composition is preferably in the form of unit dose component(s) as described above.

The present invention also provides the use of a solution comprising ascorbic acid and/or one or more salts thereof, an alkali metal or alkaline earth metal sulphate, a relatively high concentration of PEG and, optionally, further electrolytes for the treatment of patients with constipation, intestinal gas, symptoms of recurrent cramping or anorectal irritation. The PEG is consumed at a dose of more than 200g per day, preferably more than 300g per day, in divided doses. It may be provided in solid form which may be dispersed in an aqueous medium and administered from 1 to 4 times per day, preferably from 1 to 2 times per day. The number of administrations per day depends on the severity of the constipation.

EXAMPLES

Example 1. Comparison of effectiveness of 2-litre solutions of Movicol, Movicol + ascorbic acid and Movicol + ascorbic acid + sodium sulphate

6 healthy volunteers were given a 2 litre dose of each of A) Movicol, B) Movicol + ascorbic acid and C) Movicol + ascorbic acid + sodium sulphate and the volume of stools produced was measured. Movicol is a registered trademark of Norgine Limited and it is used in connection with a product of the formulation given in Table 2 below. The trial was carried out as a double blind cross-over study with 2 running periods for formulations A and B. Each volunteer was given formulation A and formulation B once each in a random order. The volunteers and the administering medical professional were blinded regarding which formulation was administered first. A third, open, study period was added for the investigation of formulation C. The composition of Movicol is shown in Table 2. The compositions of the three formulations are shown in Table 3.

Table 2: Composition per litre of Movicol

Component	Quantity
Macrogol 3350 (PEG)	105g
Sodium bicarbonate	1.428g
Sodium chloride	2.805g
Potassium chloride	0.373g
Lime and lemon flavour*	0.800g

*flavour SN292403 Lemon/Lime Nat. Trusil J2076 available from International Flavours and Fragrances (IFF)

Table 3: Compositions per litre of formulations A, B and C

Compound	Form. A	Form. B	Form. C
Movicol	1x	1x	1x
Saccharose (Vit. C placebo)	10g	0	0
Ascorbic acid	0	10g	10g
Sodium sulphate	0	0	5.6g
Osmolarity/mOsmol/kg	200	228	346

The Osmolarity values given in Table 3 are calculated values based on the information of the composition of the formulation. Movicol has a calculated osmolarity of 171 mOsmol/kg. In the calculations, polyethylene glycol is assumed to have no ionic impurities and the pH is assumed to be such that ascorbic acid essentially completely associated.

Saccharose was included in Formulation A to minimise the taste differences between the formulations. Ascorbic acid has a flavour enhancing effect on bowel preparation formulations.

Each volunteer was given 2 litres of each formulation over two hours at a rate of 250ml per 15 minutes. Stools were collected over eight hours following commencement of the treatment. The quantity of stools produced is shown in Table 4.

Table 4: Results of formulation comparison experiments

Parameter	Form. A	Form. B	Form. C
Stool weight/g	1465.2±56.7	1862±140.8	2735±199
Stool volume/l	1.4±0.0	1.8±0.1	2.7±0.2
Weight of PEG in stools/g	192.6±16.6	197.0±10.9	177.0±6.8

Table 5: Statistical significance (p) of formulation comparison experiments

Parameter	C vs A	C vs B	B vs A
Stool weight/g	<0.001	0.002	0.005
Stool volume/l	<0.001	0.003	<0.001
Weight of PEG in stools/g	0.63	0.45	0.92

As seen in Tables 4 and 5, addition of 10g/l of ascorbic acid to a Movicol formulation leads to a statistically significant increase in stool weight and stool volume.

Furthermore, addition of 10g/l of ascorbic acid and 5.6g/l of sodium sulphate to a Movicol formulation leads to an even greater statistically significant increase in stool weight and volume. Preparation C caused almost double the stool weight and volume of preparation A to be excreted. Comparing the results for preparation B and preparation C, the stool weight and volume is increased by approximately 50% (statistical significance p= 0.002 for stool weight, p=0.003 for stool volume).

The solutions containing ascorbic acid were reasonably well tolerated. Three volunteers experienced nausea whilst drinking preparation A (no ascorbic acid), whereas only two volunteers experienced nausea whilst drinking preparation B and only two volunteers experienced nausea whilst drinking preparation C. Solutions B and C were both considered to be more palatable than solution A. Solution C was, surprisingly, not considered to be less palatable than solution B. Furthermore, despite the fact that solution C was hypertonic, no adverse side effects were noted.

No other or serious adverse side effects were observed. A slight (not statistically significant) increase in blood potassium levels was recorded and a net increase in

ascorbemia was observed in the volunteers after taking solutions including ascorbic acid.

In conclusion, addition of 10g/l of ascorbic acid to a Movicol composition leads to a substantial and statistically significant increase in stool weight and stool volume. The weight and volume of stools is further dramatically increased in a statistically significant manner by the addition of 5.6g/l sodium sulphate to the Movicol/ascorbic acid composition. The solution comprising Movicol, ascorbic acid and sodium sulphate was accordingly the most effective cleansing solution, and the improved effectiveness was surprisingly not accompanied by any adverse side effects or taste compromises.

Further trials in which the properties of compositions of the present invention are compared with compositions that are currently available also show that the compositions and methods of the present invention are ones with surprisingly superior properties.

Example 2 Comparison of compositions of the invention

A trial was carried out to investigate the effect on the efficacy of the cleansing solutions of the invention of altering, independently, the quantities of the PEG, sodium sulphate and ascorbic acid components. Six compositions were investigated. The formulations were made up as aqueous solutions comprising compositions, A to F. The amount of each component in compositions A to F per litre of formulation is shown in Table 6.

Table 6: Compositions A to F of the invention

Comp.	Ingredient (g)						Osmolarity mOsmol/kg
	PEG 3350	Sodium Sulphate	Ascorbic Acid	Sodium Ascorbate	NaCl	KCl	
A	100	7.5	0.0	0.0	2.691	1.058	308
B	100	7.5	5.0	0.0	2.691	1.058	337
C	100	7.5	5.0	5.0	2.691	1.058	379
D	100	7.5	10.0	0.0	2.691	1.058	365
E	100	5.0	5.0	5.0	2.691	0.819	329
F	125	7.5	5.0	5.0	3.217	1.155	416

NB All compositions were lemon flavoured. The lemon flavouring was Ungerer Lemon SDF obtained from RSSL Pharma.

Composition E (containing 100g PEG, 5g sodium sulphate, 5g ascorbic acid, 5g sodium ascorbate, electrolytes and flavour) was the reference composition for the study.

Volunteers were informed of the aims and procedures of the study and informed consent was obtained. A medical history was obtained from each volunteer and a physical examination was carried out. 30 Volunteers were recruited. Each volunteer was randomly assigned two different treatment compositions, so that, in total, each composition was tested 10 times. The testing of each composition took one day and the two tests for each volunteer were separated by a "washout" period of 7 to 15 days.

Urine samples were collected from each volunteer throughout the day before the clinical trial. The volunteers were instructed to fast overnight before the trial and, on the day of the trial, they arrived at the clinic at 8am. Each volunteer drank 2 litres of the allocated composition over a period of 2 hours (two 125ml glasses approximately every 15 minutes). Stool volume and weight were assessed from the start of drinking and during the subsequent 8 hours. The

volunteers were generally not allowed to eat during the test period but volunteers who complained of thirst and/or appeared dehydrated were allowed to drink water 4 hours after the beginning of the study.

Weight, blood pressure and pulse rate were measured before treatment and 8 hours after the start of the treatment, or as soon as necessary in the judgement of the investigator. A blood sample was taken 4 hours after the start of the treatment for analysis of serum electrolytes, urea, creatine, hematocrit and total protein. A second blood sample was frozen for later ascorbic acid evaluation. Urine was collected between the start of the treatment and the end of the test period. Electrolytes in urine were also assessed on a volume collected during the day of the trial. Another sample of urine was frozen for later ascorbic acid evaluation.

The volunteers gave an assessment of the taste of the preparation immediately after finishing drinking the total amount of the solution.

In case of drop-out, non-compliance, or a serious adverse event not related to the study, volunteers were replaced such that 30 useful sets of data were obtained. Similarly, any volunteers who needed any medication during the trial which might influence intestinal transit or interfere with the study medication were also replaced. In total 6 volunteers had to be replaced during the study.

Each volunteer was randomly given one of the 6 compositions for the first test and a different one of the 6 compositions for the second test. Each volunteer is thus its own control and the power of the study is increased. Each PEG composition had a similar visual appearance and after dissolution in water, the volume and aspect of the compositions were similar. The compositions had different tastes.

Most of the stools were delivered 4 hours after the start of the study.

Results of the effectiveness of compositions of the invention

The volume and weights of stools collected during the study are shown in Table 7.

Table 7: Total stool volume and weight

	A	B	C	D	E	F
Volume (l)						
Mean	1.926	2.249	2.613	2.510	2.195	2.555
S.D.	0.598	0.437	0.538	0.442	0.369	0.755
Range	0.65-2.62	1.58-2.84	1.80-3.40	1.44-3.08	1.49-2.60	1.17-3.50
Weight (g)						
Mean	1992	2306	2684	2533	2283	2638
S.D.	625	437	567	442	381	745
Range	650-2744	1630-2920	1830-3557	1526-3150	1581-2765	1280-3513

The stools volume mean ranged between 1.9 and 2.6 litres.

As seen in Table 7 stool volume means for the six compositions range between 1.9 and 2.6 litres. Considering the mean values, compositions D, F and C resulted in a greater volume of stools than compositions B and E which in turn resulted in the greater volume of stools than composition A. The variability within each sample was greater than expected (average standard deviation 443ml) and consequently a global comparison between the compositions was not statistically significant ($p<0.217$). Similar results were observed for stool weight ($p<0.318$). Four individual volunteers (one each in treatments A, B, C and F) did not adhere strictly to the protocol. In the case of composition B and composition F, a small amount of vomiting took place during the treatment intake for one of the subjects, and in the case of compositions A and C, a reduced quantity of

composition solution was ingested (1000cc and 1500cc respectively) by one of the subjects. When the results were analysed excluding those treatments the statistical interpretation remained unchanged.

The time taken for the volunteers to ingest the treatment solutions was recorded and the results are shown in Table 8.

Table 8: Time for formulation intake

	A	B	C	D	E	F
Time(min)						
Mean ± S.D. Range	112.8±7.5 100-120	115.7±15.6 90-140	117.8±16.7 95-145	116.7±9.3 100-135	114.9-11.2 100-130	116.5±13.3 90-140

There was no correlation between formulation intake time and stools volume (r = -0.125 p<0.340).

Volunteers were asked to rate the taste of the solutions for salt, acid and sweetness on a scale of 0 to 3 in which 0 = very pleasant, 1 = not awkward, 2 = tolerable and 3 = intolerable. The results of the taste response are shown in Table 9.

Table 9: Taste scores

	A	B	C	D	E	F
Salt						
Mean ± S.D. Range	2.6±0.5 2-3	2.6±0.5 2-3	2.4±0.5 2-3	2.1±0.6 1-3	2.1±0.4 1-3	2.1±0.4 2-3
Acid						
Mean ± S.D. Range	1.4±0.7 0-2	2.0±0.5 1-3	1.6±0.5 1-2	1.9±0.7 1-3	1.8±0.4 1-2	1.8±0.4 1-2
Syrupy						
Mean ± S.D. Range	2.1±0.6 1-3	2.1±0.4 2-3	2.1±0.6 1-3	2.3±0.7 1-3	2.1±0 2-2	2.1±0.6 1-3

There was no significant difference between the solutions when assessed for saltiness (p<0.459) or sweetness (P<0.238). However, the assessment of acid taste of the different solutions was significantly different (p<0.039),

composition A being the least acidic and composition B the most acidic. Composition A would be expected to be least acidic as it does not contain any ascorbic acid.

In terms of efficacy there were no global statistically significant differences between the compositions. This was mostly because of the large degree of variability. However, addition of 10g ascorbic acid (ascorbic acid or mixture of ascorbic acids and sodium ascorbate) gave the best results. Treatments C and D were thus concluded to be the most effective solutions.

Clinical Laboratory Evaluation

The stools were analysed for ion contents. The results for composition D are shown in Table 10.

Table 10: Stool ionogram results for composition D

	Ionogram (mmol/litre)	Ionogram (mmol)
Na^+		
N	10	10
Mean ± S.D.	109.6	276.6±59.4
Range	97-122	148.3-375.8
K^+		
N	10	10
Mean ± S.D.	14.9±4.7	36.8±12.2
Range	8-25	24.6-64.3
Cl^-		
N	10	10
Mean ± S.D.	26.8±6.8	68.2±25.8
Range	19-42	36.0-129.4

There were no statistically significant differences between the ionograms of the six different treatments.

The hematocrit % was measured before and after treatment and the results for composition D are shown in Table 11.

Table 11: % Hematocrit before and after treatment for composition D

	% Hematocrit
Before	
N	10
Mean ± S.D.	42.1±4.4
Range	31.7-46.0
After	
N	8
Mean ± S.D.	43.3±5.6
Range	29.8-48.3
Difference	
N	9
Mean ± S.D.	1.47±0.52

Multiple comparisons between % hematocrit for the different compositions revealed no statistically significant differences. Similar results were obtained for compositions A, B, C, E and F.

Sodium, potassium, chloride and bicarbonate concentrations in the blood were measured before and after ingestion of the compositions. The results are shown in Tables 12, 13, 14 and 15.

Table 12: Change in blood sodium concentration (mmol/l)

	A	B	C	D	E	F
Before						
N	10	10	9	10	9	10
Mean ± S.D.	141.0±1.4	140.4±1.2	140.8±1.6	141.7±2.0	140.7±2.4	140.9±2.5
Range	139-143	139-142	139-143	139-145	137-144	137-145
After						
N	10	10	10	10	10	10
Mean ± S.D.	143.5±2.3	142.5±2.5	143.8±1.2	143.6±2.3	143.7±2.1	145.4±2.2
Range	139-148	137-146	142-146	140-146	140-147	141-148
Difference						
N	10	10	9	10	9	10
Mean ± S.D.	2.5±0.79	2.1±0.89	3.22±0.66	1.9±0.59	2.86±1.09	4.5±0.78

As seen in Table 12 a borderline significant difference was observed between compositions B and C ($p = 0.053$). A statistically significant difference was seen between compositions B and F ($p = 0.016$) and between compositions E

and F (p = 0.039). Composition F caused the largest increase in blood sodium levels.

Table 13: Change in blood potassium concentration (mmol/l)

	A	B	C	D	E	F
Before						
N	10	10	9	10	9	10
Mean ± S.D.	4.1±0.4	4.0±0.2	4.0±0.2	4.0±0.3	4.1±0.2	3.9±0.2
Range	3.5-4.5	3.7-4.3	3.7-4.5	3.4-4.5	3.7-4.4	3.5-4.2
After						
N	10	10	10	10	10	10
Mean ± S.D.	4.1±0.4	4.4±0.3	4.5±0.3	4.6±0.2	4.2±0.4	4.3±0.3
Range	3.3-4.6	4.0-4.9	4.0-5.0	4.2-4.9	3.5-5.0	3.9-4.8
Difference						
N	10	10	9	10	9	10
Mean ± S.D.	0.08±0.09	0.41±0.11	0.51±0.10	0.61±0.13	0.19±0.14	0.43±0.07

Multiple comparisons showed no significant difference between the compositions.

Table 14: Change in blood chloride concentration (mmol/l)

	A	B	C	D	E	F
Before						
N	10	10	9	10	9	10
Mean ± S.D.	102.3±1.6	101.4±2.5	102.1±1.4	102.5±2.3	102.3±1.7	103.3±2.6
Range	100-105	98-105	100-104	100-107	100-105	100-107
After						
N	10	10	10	10	10	10
Mean ± S.D.	105.6±1.8	106.2±3.2	106.4±2.4	107.4±2.4	105.7±1.5	108.9±3.3
Range	103-108	102-112	104-112	104-111	104-108	105-114
Difference						
N	10	10	9	10	9	10
Mean ± S.D.	3.3±0.80	4.8±0.87	4.44±0.80	4.9±0.41	3.22±0.62	5.6±0.69

As seen in Table 14 a difference of border line statistical significance was observed between compositions A and D (p = 0.056). A significant difference was observed between compositions A and F (p = 0.010), compositions B and F (p = 1.036), compositions D and E (p = 0.031) and compositions E and F (p = 0.005). Composition F caused the largest increase in blood chloride concentration.

Table 15: Change in blood bicarbonate concentration (mmol/l)

	A	B	C	D	E	F
Before						
N	10	10	9	10	9	10
Mean ± S.D.	26.9±3.3	28.2±2.0	28.6±2.1	28.9±2.0	28.4±2.1	26.3±3.2
Range	19-30	25-31	26-32	26-32	25-32	20-31
After						
N	10	10	10	10	10	10
Mean ± S.D.	26.7±3.3	26.1±2.1	26.6±1.9	25.5±1.8	26.1±2.2	25.6±2.5
Range	25-29	24-31	24-30	22-27	22-30	23-31
Difference						
N	10	10	9	10	9	10
Mean ± S.D.	-0.2±0.80	-2.1±0.82	-1.78±0.60	-3.4±0.60	-1.89±0.72	-0.7±0.75

Multiple comparisons showed a significant difference between compositions A and D (p = 0.010) and between compositions E and F (p = 0.035)

Blood urea, creatinemia and protidemia were measured. The results for composition D are shown in Table 16.

Table 16: Change in Blood urea, creatinemia and protidemia for composition D

	Urea (mmol/l)	Creatinemia (mmol/litre)	Protidemia (g/l)
Before			
N	10	10	10
Mean ± S.D.	4.9±0.9	81.4±11.9	75.3±4.8
Range	3.6-6.3	64-98	69-83
After			
N	10	10	10
Mean ± S.D.	4.5±1.0	80.9±13.6	78.9±5.8
Range	2.6-5.8	59-98	70-87
Difference			
N	10	10	10
Mean ± S.D.	-0.39±0.21	-0.50±1.66	3.60±1.77

Multiple comparisons between the compositions showed no significant differences. Similar results were obtained for compositions A, B, C, E and F.

Multivariate analysis of differences for all biological parameters showed no significant results.

Urine was also analysed for sodium, potassium and chloride content. The results for the group of volunteers given composition D are shown in Table 17.

Table 17: Change in urine sodium, potassium and chloride content for composition D

	Sodium Content (mmol)	Potassium content (mmol)	Chloride content (mmol)
Before			
N	10	10	9
Mean ± S.D.	63.6±33.8	38.9±21.3	69.8±34.6
Range	23.5-105.0	7.8-69.0	19.7-112.0
After			
N	10	10	10
Mean ± S.D.	46.6±47.1	24.8±20.8	53.7±48.4
Range	3.1-161.6	4.9-64.8	6.8-164.8

A slight, not statistically significant, decrease in the level of urine sodium, potassium and chloride was observed. There was no statistically significant difference between the composition treatment groups. Similar results were obtained for compositions A, B, C, E and F.

Ascorburia was also measured and the results are shown in Tables 18 and 19.

Table 18: Ascorburia (μmol/litre)

	A	B	C	D	E	F
N	8	8	7	7	4	4
Mean	445.3	5266.9	7292.1	8408.6	8046.3	2556.5
S.D.	668.3	4402.9	1781.4	9641.3	2437.7	3690.4
Range	17-1756	158-15141	4913-9463	129-28390	4482-9835	338-8043

Table 19: Ascorburia (μmol)

	A	B	C	D	E	F
N	8	8	7	7	4	4
Mean	154.3	1690.1	2631.3	2152.1	2423.6	1141.2
S.D.	223.0	1473.1	1134.9	1856.4	1406.1	1818.1
Range	2.5-527	47-4845	1103-4216	77-5678	448-3740	108-3861

There was no significant differences between compositions in terms of μmol/lire or μmol ascorburia (pc 0.303 and pc 0.641 respectively). As expected, composition A

showed the lowest level of ascorburia because the solution did not contain any ascorbic acid or sodium salts.

In conclusion, all biochemical alterations were without clinical significance and all compositions were clinically and biologically well tolerated.

Conclusions of experiments directed to compositions of the invention:

Table 20: Summary of results for compositions A to F

Comp	Ingredient (g)				Osmol. mOsmol/kg	Mean Stool Volume l	Mean Decrease in HCO_3^- mmol/l
	PEG 3350	Sodium Sulphate	Ascorbic Acid	Sodium Ascorbate			
A	100	7.5	0.0	0.0	308	1.9	0.2
B	100	7.5	5.0	0.0	337	2.2	2.1
C	100	7.5	5.0	5.0	379	2.6	1.8
D	100	7.5	10.0	0.0	365	2.5	3.4
E	100	5.0	5.0	5.0	329	2.2	1.9
F	125	7.5	5.0	5.0	416	2.6	0.7

As is seen from the data in Table 20, compositions comprising 7.5g sodium sulphate (A, B, C, D and F) gave rise to larger volume of stools than the composition comprising only 5g sodium sulphate (E). No significant difference was observed between equivalent solutions containing 100g polyethylene glycol (C) and 125g polyethylene glycol (F). Compositions containing ascorbic acid and/or sodium ascorbate (B, C, D, E and F) gave rise to larger volumes of stools than the composition without ascorbic acid or sodium ascorbate (A). Compositions containing 10g ascorbic acid (D) or 5g ascorbic acid plus 5g sodium ascorbate (C, E and F) gave rise to a larger volume of stools than compositions containing 5g ascorbic acid alone (B). The preparation containing 100g polyethylene glycol, 7.5g sodium sulphate and 10g ascorbic

acid induced a clinically significant fall in plasma bicarbonate levels (D). That fall was not observed in the case of the composition containing 100g polyethylene glycol, 7.5g sodium sulphate, 5g ascorbic acid and 5g sodium ascorbate (C).

Weight loss was around 1kg for all volunteers despite a decrease in urinary volume (200-300 ml over 8 hours). Protidemia and haematocrit were increased explaining a slight dehydration. Natremia and kaliemia were also slightly increased. Polyethylene glycol concentration in stools was assessed in treatment groups D and F only. Stool volumes correlated roughly to the amount of PEG measured in the collected stools.

Example 3: Study of ingestion of colon cleansing composition interspersed with water ingestion

Ten subjects were enrolled for the study and written informed consent was given before its commencement. Each volunteer was given 2 litres of composition D as defined in Example 2. The composition was administered according to two different modes of administration on two different occasions separated by a wash-out period. In administration mode 1, the volunteer drank 2 litres of composition D within 2 hours as in the case of Example 2. According to administration mode 2 the volunteer drank 1 litre of composition over 1 hour, followed by 500 ml water over the next hour, followed by a second litre of composition over the following hour, followed by 1000 ml water over the following 2 hours.

Stools were collected over the 8 hours following commencement of the administration. A comparison of stool weights obtained by the two modes of administration is shown in Table 21.

Table 21: Comparison of stool weight according to mode of administration

	Mode 1	Mode 2
Weight (g)		
N*	9	9
Mean	2464	2726
S.D.	409	198
Range	1526-2865	2350-2920

* Number of subjects

The volume of stools generated following administration of composition D is increased by around 300 ml using mode 2 as compared with mode 1. When asked to rate their impression of the tolerability of the treatment on a visual analogue scale (VAS) on which 0mm = excellent appreciation and 100mm = very bad appreciation the tolerability was ranked as shown in Table 22.

Table 22: Comparison of tolerability VAS according to mode of administration

	Mode 1	Mode 2
VAS (mm)		
N	9	9
Mean ± S.D.	68.4±20.0	59.4±21.0
Range	35-98	17-85

A statistically significant decrease ($p<0.0276$) of 10mm on the VAS scale in favour of better tolerability for mode 2 of administration was observed. Among nine subjects, eight preferred the second mode of administration.

Example 4 Assessment of efficacy and safety of colon cleansing solution in patients undergoing endoscopy

30 patients (12 male, 18 female, mean age 51 +/- 11) were given 2 litres of a colon cleansing composition comprising for one litre of solution the materials as shown in table 23

Table 23 Endoscopy experiment colon cleansing solution

Material	Weight (g) per 1 preparation
PEG 3350	100.0
Sodium Sulphate	7.5
Ascorbic Acid	4.7
Sodium Ascorbate	5.9
Sodium Chloride	2.69
Potassium Chloride	0.93
Lemon Flavour	2.015
Citric Acid Anhydrous	1.565
Acesulfame K	0.35
Calculated osmolarity	392 mOsmol/kg

The colon cleansing solution was given in a regime as follows:

First hour: 1 litre of solution ingested orally

Second hour: 0.5 litre water ingested orally

Third hour: 1 litre of solution ingested orally

Fourth hour: at least 0.5 litre water ingested orally within 1.5 hours

The cleanliness of the colon was assessed by the colonoscopist on a 4 point scale (very good for all colon segments = grade 4, good for all colon segments = grade 3, at least one colon segment with totally or partially removable residual faeces = grade 2 or 1, at least one colon segment with heavy hard stools = grade 0). The scores were then

ranked as A for grade 3 or 4, B for grade 1 or 2 and C for grade 0. The investigator judged the quality of colon preparation as very good or good in 20 patients, in at least one section moderate in 6, bad in 3 and very bad in 1, leading to a final scoring of 20A, 9B and 1C.

The mean weight of stools was 2866 +/- 667 g and the mean volume of fluid removed from the colon during colonoscopy was 130 +/- 124 ml. The digestive tolerance of the preparation was good in 26 patients, moderate in 2 and poor in 2. Only one patient experienced profuse vomiting. No statistically significant changes in blood chloride or bicarbonate ion concentrations were observed over the period of the treatment.

Example 5 Formulation examples

Flavoured Product Formulation 1

Table 23 UNGERER LEMON Composition

Material	% w/w	Weight (g) per 125ml preparation
Movicol Base	96.275	13.7008
Acesulfame K	0.321	0.0455
Talin	0.058	0.0082
NHDC	0.058	0.0082
Citric Acid	0.078	0.0110
Natrosol 250 M	1.840	0.2606
Lemon	0.920	0.1303

Talin is a taste enhancer comprising Thaumatin (available from The Talin Food Company, Merseyide, England). NHDC (neohesperidine dihydrochloride) is a sweetener, (available from Evesa, P.O. Box 103, 11300 La Linea de la concepcion,

Cádiz, Spain). Natrosol 250M is a hydroxyethylcellulose available from Hercules Incorporated via Aqualon.

Flavoured Product Formulation 2

Table 24 UNGERER STRAWBERRY Composition Product

Material	% w/w	Weight (g) per 125ml preparation
Movicol Base	97.603	13.7008
Acesulfame K	0.325	0.0456
NHDC	0.014	0.0020
Natrosol 250 M	1.858	0.2608
Strawberry	1.800	0.2527

Flavoured Product Formulation 3

Table 25 IFF Grapefruit flavouring Composition

Material	% w/w	Weight (g) per 2l preparation
PEG 3350	79.24	200.0
Sodium Sulphate	5.94	15.0
Ascorbic Acid	3.96	10.0
Sodium Ascorbate	3.96	10.0
Sodium Chloride	2.13	5.38
Potassium Chloride	0.84	2.12
IFF Grapefruit Flavouring Powder	2.41	6.08
Citric Acid Anhydrous	1.23	3.10
Acesulfame K	0.28	0.70
Calculated osmolarity	392 mOsmol/kg	392 mOsmol/kg

Flavoured Product Formulation 4

Table 26 UNGERER BLACKCURRANT Composition

Material	% w/w	Weight (g) per 21 preparation
PEG 3350	79.81	200.0
Sodium Sulphate	5.99	15.0
Ascorbic Acid	3.99	10.0
Sodium Ascorbate	3.99	10.0
Sodium Chloride	2.15	5.38
Potassium Chloride	0.85	2.12
Ungerer Blackcurrant	1.62	4.06
Citric Acid Anhydrous	1.24	3.10
Acesulfame K	0.28	0.70
Talin	0.08	0.20
Calculated osmolarity	392 mOsmol/kg	392 mOsmol/kg

Flavoured Product Formulation 5

Table 27 IFF Pineapple flavouring

Material	% w/w	Weight (g) per 21 preparation
PEG 3350	79.81	200.0
Sodium Sulphate	5.99	15.0
Ascorbic Acid	3.99	10.0
Sodium Ascorbate	3.99	10.0
Sodium Chloride	2.15	5.38
Potassium Chloride	0.85	2.12
IFF Pineapple Flavouring Powder	1.70	4.06
Citric Acid Anhydrous	1.26	3.10
Acesulfame K	0.29	0.70
Calculated osmolarity	392 mOsmol/kg	392 mOsmol/kg

Flavoured Product Formulation 6

Table 28 IFF Vanilla + Givaudan-Roure Lemon and Lime Flav-o-lok Composition

Material	% w/w	Weight (g) per 2l preparation
PEG 3350	78.95	200.0
Sodium Sulphate	5.92	15.0
Ascorbic Acid	3.95	10.0
Sodium Ascorbate	3.95	10.0
Sodium Chloride	2.12	5.38
Potassium Chloride	0.84	2.12
IFF Vanilla Flavouring Powder	1.61	4.07
Givaudan-Roure Lemon and Lime Flav-o-lock	1.15	2.91
Citric Acid Anhydrous	1.22	3.09
Acesulfame K	0.29	0.70
Calculated osmolarity	392 mOsmol/kg	392 mOsmol/kg

In the following examples, the ascorbic acid and sodium ascorbate components are packaged separately from the other components to improve their stability.

a) Citric acid-containing composition

The composition is provided in two sachets. The contents of the two sachets together are for making up to one litre of colon cleansing solution by addition of water.

SACHET 1

PEG 3350:	100.000 g
Sodium Sulphate:	7.500 g
Sodium Chloride:	2.691 g
Potassium Chloride:	0.930 g
Anhydrous Citric Acid:	1.565 g
Acesulfame K:	0.350 g
Lemon Flavour:	2.015 g
TOTAL WEIGHT OF SACHET 1 INGREDIENTS	115.051 g

SACHET 2

Ascorbic Acid:	4.700 g
Sodium Ascorbate:	5.900 g
TOTAL WEIGHT OF SACHET 2 INGREDIENTS	10.600 g

b) Aspartame-containing composition

The composition is provided in two sachets. The contents of the two sachets together are for making up to one litre of colon cleansing solution by addition of water.

SACHET 1 (in grams per litre)

PEG 3350:	100.000 g
Sodium Sulphate:	7.500 g
Sodium Chloride:	2.691 g
Potassium Chloride:	1.015 g
Aspartame:	0.233 g
Acesulfame K:	0.117 g
Lemon Flavour (Ungerer V3938-1N1)	0.340 g
TOTAL WEIGHT OF SACHET 1 INGREDIENTS	111.896 g

SACHET 2 (in grams per litre)

Ascorbic Acid:	4.700 g
Sodium Ascorbate:	5.900 g
TOTAL WEIGHT OF SACHET 2 INGREDIENTS	10.600 g

Comparative Examples

Various investigations were made into the reduction of the volume of solutions of the prior art required to achieve satisfactory colon clearance. Stimulant laxatives, for example bisacodyl, picosulphate or senna, were added to a GoLYTELY solution. They were effective in achieving clearance of the bowel, but the risk of plasma electrolyte disturbances was increased. In a further experiment, a hyper-concentrated GoLYTELY solution (powder for two one-litre doses made up to only one litre) was found to be effective in the clearance of the bowel, but the solution was unpalatable. That is to say participants in the trial found the taste of the solution so unpleasant that ingestion of the solution was very difficult. In the absence of direct supervision by a medical professional, that degree of unpalatablity is likely to lead to patient non-compliance.

Claims

1. A dry composition for admixture with water wherein the dry composition comprises, per litre of aqueous solution to be made, the following components:

a) 80 to 350g polyethylene glycol

b) 3 to 20g ascorbic acid, one or more salts of ascorbic acid or a mixture of ascorbic acid and one or more salts of ascorbic acid

c) 1 to 15g of an alkali metal or alkaline earth metal sulphate or a mixture of alkali metal or alkaline earth metal sulphates; and

d) optionally one or more electrolytes selected from sodium chloride, potassium chloride and sodium hydrogen carbonate

the components of the composition being selected such that an aqueous solution made up to 1 litre has an osmolarity within the range of from 300 to 700 mOsmol/kg.

2. A dry composition as claimed in claim 1 wherein an aqueous solution made up in 1 litre of water has an osmolarity of 330 mOsmol/kg or greater.

3. A dry composition as claimed in claim 1 or claim 2 wherein an aqueous solution made up in 1 litre of water has an osmolarity in the range 550 mOsmol/kg or lower.

4. A dry composition as claimed in any one of claims 1 to 3 wherein the polyethylene glycol has an average molecular weight of 2000 or greater.

5. A dry composition as claimed in any one of claims 1 to 4 wherein the polyethylene glycol has an average molecular weight of 4500 or lower.

6. A dry composition as claimed in any one of claims 1 to 5 wherein the polyethylene glycol has an average molecular weight of 3350 or 4000.

7. A dry composition as claimed in any one of claims 1 to 6 comprising from 90g or more of polyethylene glycol per litre.

8. A dry composition as claimed in any one of claims 1 to 7 comprising from 150g or less of polyethylene glycol per litre.

9. A dry composition as claimed in any one of claims 1 to 8 comprising 100g of polyethylene glycol per litre.

10. A dry composition as claimed in any one of claims 1 to 9 comprising the alkali metal or alkaline earth metal sulphate or the mixture of alkali metal or alkaline earth metal sulphates in a quantity of 3g or more per litre.

11. A dry composition as claimed in any one of claims 1 to 10 comprising the alkali metal or alkaline earth metal sulphate or the mixture of alkali metal or alkaline earth metal sulphates in a quantity 9g or less per litre.

12. A dry composition as claimed in claim 10 or 11 comprising the alkali metal or alkaline earth metal sulphate or the mixture of alkali metal or alkaline earth metal sulphates in a quantity of 7.5g per litre.

13. A dry composition as claimed in any one of claims 1 to 12 wherein the alkali metal or alkaline earth metal is magnesium.

14. A dry composition as claimed in any one of claims 1 to 12 wherein the alkali metal or alkaline earth metal is sodium.

15. A dry composition as claimed in any one of claims 1 to 14 which comprises sodium chloride.

16. A dry composition as claimed in claim 15 comprising sodium chloride in a quantity of 2g or more per litre.

17. A dry composition as claimed in any one of claims 1 to 16 which comprises potassium chloride.

18. A dry composition as claimed in claim 17 comprising potassium chloride in a quantity of from 0.5g or more per litre.

19. A dry composition as claimed in any one of claims 1 to 18 comprising ascorbic acid or a salt thereof in a quantity of 5g or more per litre.

20. A dry composition as claimed in any one of claims 1 to 19 comprising ascorbic acid or a salt thereof in a quantity of 10g or less per litre.

21. A dry composition as claimed in any one of claims 1 to 20 comprising ascorbic acid and a salt thereof.

22. A dry composition as claimed in claim 21 comprising ascorbic acid and a salt thereof in a weight ratio within the range of from 1:9 to 9:1.

23. A dry composition as claimed in any one of claims 1 to 22 comprising sodium ascorbate.

24. A dry composition as claimed in any one of claims 1 to 23 further comprising a flavouring.

25. A dry composition as claimed in claim 24 wherein the flavouring is lemon.

26. A dry composition as claimed in any one of claims 1 to 24 further comprising a sweetener which is not a substrate for bacteria in the gut.

27. A dry composition as claimed in claim 26 wherein the sweetener is selected from aspartame, acesulfame K, saccharine and citric acid.

28. A dry composition as claimed in any one of claims 1 to 27 in which the ascorbic acid and/or the salt of ascorbic acid is coated.

29. A colon cleansing preparation comprising an aqueous solution of a composition as claimed in any one of claims 1 to 28.

30. A colon cleansing preparation sachet comprising the components of a composition as claimed in any one of claims 1 to 28.

31. A colon cleansing composition kit comprising the components of a composition as claimed in any one of claims 1

to 28, comprising at least two portions in which the ascorbic acid and/or the salts thereof are packaged in one portion and the other components are packaged in the other portion.

32. A method of cleansing the colon of a mammal, comprising administering orally to the mammal a cleansing fluid comprising, per litre, the following components:

a) 80 to 150g polyethylene glycol;

b) 3 to 20g ascorbic acid, one or more salts of ascorbic acid or a mixture of ascorbic acid and one or more salts of ascorbic acid;

c) 1 to 15g of an alkali metal or alkaline earth metal sulphate or a mixture of alkali metal or alkaline earth metal sulphates; and

d) optionally one or more electrolytes selected from sodium chloride, potassium chloride and sodium hydrogen carbonate

the components being selected such that the cleansing fluid has an osmolarity within the range of from 300 to 700 mOsmol/kg, the volume of fluid administered being from 1.5 to 3 litres for an adult human and pro rata for a mammal other than an adult human.

33. A method as claimed in claim 32 wherein the composition is administered over 1 to 4 hours.

34. A method as claimed in claim 32 or claim 33 wherein the colon is cleansed prior to carrying out a diagnostic, therapeutic or surgical procedure on the colon, rectum or anus or elsewhere in the abdomen.

35. A method as claimed in any one of claims 32 to 34 wherein the cleansing fluid used in the method has the features of the compositions described in any one of claims 1 to 29.

36. A dry composition for admixture with water wherein the dry composition comprises, per litre of aqueous solution to be made, the following components:

- 30 to 350 g polyethylene glycol

- 3 to 20g ascorbic acid and one or more salts of ascorbic acid
- optionally one or more electrolytes selected from sodium chloride, potassium chloride, sodium hydrogen carbonate and the alkali metal or alkaline earth metal sulphates.

37. A dry composition as claimed in claim 36 wherein the salt of ascorbic acid is an alkali metal or alkaline earth metal ascorbate.

38. A dry composition as claimed in claim 36 wherein the salt of ascorbic acid is sodium ascorbate.

39. A dry composition as claimed in any one of claims 36 to 38 wherein the ascorbic acid and the salt thereof are present in a weight ratio within the range of from 1:9 to 9:1.

40. A dry composition as claimed in any one of claims 36 to 39 wherein the ascorbic acid and the salt thereof are present in a weight ratio within the range of from 4:6 to 6:4.

41. A dry composition as claimed in any one of claims 36 to 40 wherein the composition comprises the additional features described in claims 4 to 28.

42. A dry composition as claimed in any one of claims 36 to 41 wherein the composition comprises sodium sulphate.

43. A colon cleansing preparation comprising an aqueous solution of a composition as claimed in any one of claims 36 to 42.

44. A colon cleansing preparation sachet comprising the components of a composition as claimed in any one of claims 36 to 42.

45. A colon cleansing composition kit comprising the components of a composition as claimed in any one of claims 36 to 42, comprising at least two portions in which the ascorbic acid and the salt thereof are packaged in one portion and the other components are packaged in the other portion.

46. A method for cleansing the colon of a mammal comprising administering orally to the mammal a preparation comprising, per litre, the following components:

- 30 to 350 g polyethylene glycol
- 3 to 20 g of a mixture of ascorbic acid and a salt of ascorbic acid
- optionally one or more electrolytes selected from sodium chloride, potassium chloride, sodium hydrogen carbonate and the alkali metal or alkaline earth metal sulphates.

the volume of composition administered being from 1.5 to 4 litres for an adult human and pro rata for a mammal other than an adult human.

47. A method as claimed in claim 46 wherein the preparation is administered over 1 to 4 hours.

48. A method as claimed in claim 46 or 47 wherein the colon is cleansed prior to carrying out a diagnostic, therapeutic or surgical procedure on the colon, rectum or anus or elsewhere in the abdomen.

49. A method as claimed in any one of claims 46 to 48 wherein the composition used in the method is as described in any one of claims 36 to 42.

50. The use of a PEG for the manufacture of a medicament for cleansing the colon of a mammal according to an administration regime comprising the consecutive steps of

a) administering 0.5 up to 3.0 litres of a colon cleansing solution comprising PEG (volume V_{PEG}) over a period of time t_1 and

b) administering 0.3 up to 2.0 litres of clear fluid (volume V_{cf}) over a period of time t_2.

51. The use as claimed in claim 50 wherein t_1 is 15 minutes or more.

52. The use as claimed in claim 50 or 51 wherein t_1 is 2 hours or less.

53. The use as claimed in any one of claims 50 to 52 wherein t_2 is 15 minutes or more.

54. The use as claimed in any one of claims 50 to 53 wherein t_2 is to 2 hours or less.

55. The use as claimed in any one of claims 50 to 54 wherein V_{PEG} is 500ml or greater.

56. The use as claimed in any one of claims 50 to 55 wherein V_{PEG} is 2000ml or less.

57. The use as claimed in any one of claims 50 to 56 wherein V_{cf} is 300ml or greater.

58. The use as claimed in any one of claims 50 to 57 wherein V_{cf} is 1500ml or less.

59. The use as claimed in any one of claims 50 to 58 wherein the PEG solution comprises a composition as described in any one of claims 1 to 28 or 36 to 42.

60. A method of cleansing the colon of a mammal, comprising administering orally to the mammal in consecutive steps:

a) 0.5 up to 3.0 litres of a cleansing solution comprising PEG (volume V_{PEG}) over a period of time t_1 and

b) 0.3 up to 2.0 litres of clear fluid (volume V_{cf}) over a period of time t_2.

61. A method as claimed in claim 60 wherein t_1 is 15 minutes or more.

62. A method as claimed in claim 60 or claim 61 wherein t_1 is 2 hours or less.

63. A method as claimed in any one of claims 60 to 62 wherein t_2 is 15 minutes or more.

64. A method as claimed in any one of claims 60 to 63 wherein t_2 is 2 hours or less.

65. A method as claimed in any one of claims 60 to 64 wherein V_{PEG} is 800ml or greater.

66. A method as claimed in any one of claims 60 to 65 wherein V_{PEG} is 2000ml or less.

67. A method as claimed in any one of claims 60 to 66 wherein V_{cf} is 400ml or greater.

68. A method as claimed in any one of claims 60 to 66 wherein V_{cf} is 1500ml or less.

69. Use of a PEG for the manufacture of a medicament for cleansing the colon of a patient according to an administration regime comprising the consecutive steps of:

a) administering 0.5 up to 3.0 litres of a first PEG-containing colon cleansing solution (volume V_{PEG}) over a period of time t_1

b) administering 0.3 up to 2.0 litres of clear fluid (volume V_{cf}) over a period of time t_2, and

c) administering 0.5 up to 3.0 litres of a second PEG-containing colon cleansing solution (volume $V2_{PEG}$) over a period of time t_3.

70. The use as claimed in claim 69 wherein V_{PEG}, t_1, V_{water} and t_2 are as described in any one of claims 51 to 58.

71. The use as claimed in claim 68 or 69 wherein t_3 is approximately one hour.

72. The use as claimed in any one of claims 69 to 71 wherein $V2_{PEG}$ is 800ml or greater.

73. The use as claimed in any one of claims 69 to 72 wherein $V2_{PEG}$ is 2000ml or less.

74. The use as claimed in any one of claims 68 to 73 wherein the administration regime comprises the further step

d) administering 0.5 up to 2.0 litres of water (volume $V2_{water}$) over a period of time t_4.

75. The use as claimed in claim 74 wherein $V2_{water}$ is 800ml or greater.

76. The use as claimed in claim 74 or 75 wherein $V2_{water}$ is 1500ml or less.

77. The use as claimed in any one of claims 74 to 76 wherein t_4 is 30 minutes or more.

78. The use as claimed in any one of claims 74 to 77 wherein t_4 is 3 hours or less.

79. The use as claimed in any one of claims 69 to 78 wherein the PEG solution comprises a composition as described in any one of claims 1 to 28 or 36 to 42.

80. A unit dosage colon cleansing composition comprising two components, the first component being a composition for making up a first PEG-containing colon cleansing solution and the second component being a composition for making up a second PEG-containing colon cleansing solution.

81. A unit dosage colon cleansing composition as claimed in claim 80 wherein the kit comprises at least one further component comprising ascorbic acid and/or one or more salts thereof packaged separately from other components, said further component being for admixture with the first or second composition for making up a PEG-containing colon cleansing solution.

82. A kit comprising a box, the compositions a unit dosage colon cleansing composition as claimed in claim 80 or 81 and instructions for its use.

83. A method as claimed in any one of claims 32, 33 and 46 to 48 wherein the colon is cleansed in the treatment of an acute gastrointestinal infection.